SOLUTE EFFECTS OF BORON AND ZIRCONIUM ON MICROPOROSITY

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ABSTRACT

It is generally agreed that solutes which tend to increase the solidification range of an alloy significantly affect microporosity in castings. Binary phase diagrams suggest that boron and zirconium - solutes which are traditionally added in small amounts to cast nickel-base alloys - might significantly affect the solidification range, hence microporosity, of cast nickel-base superalloys. The effects of these suspect solutes on the solidification range and mechanical properties of a highly alloyed nickel-base superalloy were measured and correlated with the microporosity levels in production castings produced from these alloys. Based on the results of this study it would seem that: boron content should be maintained at the traditional 0.01% level; and zirconium should be eliminated from nickel-base alloys used for microporosity-limited applications.

Introduction

Microporosity in conventional castings is a common microdefect emanating from gas exsolution and/or solidification shrinkage. Gas-induced microporosity is not normally regarded as a problem with nickel-base superalloys because of their high "reactive" element content but shrinkage-induced microporosity is virtually inevitable with nickel-base superalloys.

Shrinkage-induced microporosity is thought to occur when the local liquid metal pressure (P₁) ceases to be positive during the latter stages of solidification. Liquid metal pressures during the initial stages of solidification are generally high and correspond to a first approximation to the sum of the ambient pressure (P_a) and the metallostatic pressure (P_h), i.e. $P_1 = P_a + P_h$. As solidification progresses, P_1 continually decreases due to the pressure drop (ΔP) associated with feeding liquid metal through a narrow flow channel. (1,2,3)

Nickel-base superalloys are particularly susceptible to shrinkage-induced microporosity since most nickel-base superalloys are cast in vacuum; therefore, initial liquid metal pressures do not benefit from ambient pressure. In addition, nickel-base superalloys tend to be "mushy freezing" and, to accommodate solidification shrinkage, feeding must be accomplished through narrow interdendritic channels causing relatively large pressure drops during the latter stages of solidification.

Several attempts have been made to calculate ΔP based on fundamental fluid flow and heat transfer considerations. Proposed solutions vary, but they generally indicate the pressure drop is least when the local solidification time (t_f) is greatest and/or when the alloy solidification range (ΔT), i.e. difference between alloy liquidus and solidus temperatures, is least. To a first approximation, the relationship between pressure drop, solidification range and solidification time can be expressed as:

$$\Delta P = \frac{k(\Delta T)^2}{t_f}$$

where k is a constant of proportionality.

Indeed, many foundry "tricks of the trade" to minimize microporosity in nickel-base superalloy castings simply involve increasing ambient pressure (quick release from vacuum, hot top, etc.) or increasing local solidification time (high mold temperatures, low thermal conductivity shell materials, etc.). However, very few if any practical attempts have been made to minimize microporosity by manipulating the alloy solidification range.

"Melting ranges" for cast nickel-base superalloys reportedly vary between 27°C (IN 713C) to 72°C (IN 100) with an average of about 42°C. Reducing the melting or solidification range by as little as 5°C could conceivably reduce microporosity by as much as 25%!(4,5)

The solidification range for unalloyed nickel is nil but increases with alloying additions up to the maximum solid solubility of the alloying additions. Alloying additions in excess of the solid solubility limit reduce the solidification range to nil at the eutectic. The effect of individual alloy additions on the solidification range of a nickel-base superalloy can be estimated, albeit imprecisely, from binary phase diagrams. Consider for example the nickel-aluminum system illustrated in Fig. 1. The maximum solubility of aluminum in nickel is approximately 9.6% (by weight). The equilibrium solidus and liquidus temperatures at 9.6% aluminum are approximately 1402°C and 1385°C, respectively, corresponding to a solidification range of about 17°C at 9.6% Al, i.e. ~2°C per weight percent aluminum. Since aluminum is seldom added to nickel-base superalloys in amounts greater than



Fig. 1. Nickel-rich portion of the Fig. 2. Gas turbine integral wheel nickel-aluminum phase diagram. investment casting.

the solid solubility limit, it is reasonable to estimate that aluminum will increase the solidification range of nickel-base superalloys by ~2°C per weight percent aluminum.(6)

Obviously, major modifications in aluminum content are required to significantly affect the solidification range of nickel-base superalloys. Solutes such as boron and zirconium could have a much greater effect on the solidification range of nickel-base superalloys.

The maximum solid solubility of boron in nickel is 0.015% by weight. Boron additions less than the solubility limit moderately affect the binary liquidus temperatures while significantly lowering binary solidus temperatures. The solidification range at 0.015% boron is greater than 300°C indicating that boron could potentially increase the solidification range of nickel-base superalloys by as much as 2500°C per weight percent boron. The solid solubility of boron in actual nickel-base superalloys must, however, be considerably less than 0.015% since the solidification range of actual nickel-base superalloys containing as much as 0.015% boron is only of the order of 40 to 45°C. Nevertheless, boron remains an interesting solute with respect to its effects on the solidification range and microporosity.

Zirconium is surprisingly similar to boron with respect to its effect on binary alloy solidification ranges. Zirconium in amounts less than the binary solid solubility limit (0.9%) moderately affects liquidus temperatures but significantly lowers solidus temperatures. The solidification range at 0.9% zirconium is ~260°C indicating that zirconium could increase the solidification range of nickel-base superalloys by as much as 280°C per weight percent zirconium.

While boron and zirconium are normally added to nickel-base superalloys in amounts ranging between 0.005% to 0.020% and 0.05% to 0.15%, respectively, their effects on solidification range - even in trace amounts - are potentially significant.

Experimental

The actual effects of trace additions of boron and zirconium on the solidification range and microporosity level in nickel-base superalloys were experimentally assessed by producing castings from a nickel-base superalloy in which boron and zirconium were intentionally varied. The base alloy selected for study was a versatile Ni-10Co-9Cr-9W-5.5Al-3.5Ta-1.5Mo-1.5Ti superalloy developed by Ford Motor Company especially for gas turbine integral wheel applications. Experimental modifications of the base alloy were prepared by vacuum induction melting a master heat to which controlled amounts of zirconium and/or boron were subsequently added. The casting practices used to produce the castings were representative of the "state of the art" and the cast wheel configuration is shown in Fig. 2.

The castings were sectioned and examined using conventional metallographic techniques. Microporosity levels in critical disc/rim sections were carefully examined at 50X to precisely identify the most porous area in that section. The areal percentage of microporosity was measured directly from 3-1/2" x 4-1/2" photographs of the selected areas. In effect, the microporosity measurements in this investigation properly refer to the microporosity level in the worst 0.0063 square inch field of view.

Microstructural studies were carried out on samples which were electropolished only as well as in the etched condition. Structural features were recorded with a JEOLCO SEM and phases in situ were chemically analyzed with a PGT 1000 nondispersive analyzer. X-ray diffraction studies of extracted residues were carried out.

The solidification range of each alloy was measured using a DuPont Model 900 Differential Thermal Analyzer (DTA) equipped with an unalloyed nickel reference cell. The DTA samples, in the form of filings, were taken directly from the hub sections of the castings. The DTA samples were heated at ~15°C per minute in a flowing (~45 liters per hour) argon atmosphere to a maximum temperature of ~1500°C and free-cooled back to room temperature. A typical DTA thermogram for base alloy remelt bar is illustrated in Fig. 3.



Fig. 3. Typical DTA thermogram for Ni-10Co-9Cr-9W-5.5Al-3.5Ta-1.5Mo-1.5Ti

Solidus and liquidus temperatures were measured during heating and cooling. The solidus temperature during heating was interpreted as the temperature of tangency between the baseline and the endothermic peak denoting melting. The liquidus temperature was interpreted as the endothermic peak inflection temperature. Reverse procedures were used to determine the liquidus and solidus temperatures during cooling. All measured temperatures were calibrated and corrected to the melting point of the unalloyed nickel reference which was assumed to be 1453°C.

Mechanical test specimens were machined from the disc sections of the integral wheel castings and were tested in stress-rupture at 760° C, 60 kg/mm^2 (1400°F, 85 ksi) and 928°C, 30 kg/mm² (1700°F, 42 ksi). The specimens were carefully machined from the same relative circumferential locations in the castings to minimize possible solidification kinetic effects.

Results

Zirconium Effects

The effects of zirconium were determined at three zirconium levels: nil, 0.07% and 0.15% while the boron level was maintained at 0.012%. Solidification range, microporosity and mechanical property measurements at the three zirconium levels are summarized in Table 1.

Zirconium increased the solidification range of the base alloy from 68° C without zirconium to 84° C at 0.15% zirconium. The effect of zirconium on Δ T was not as great as anticipated from binary phase diagram considerations but was significant, nevertheless. Apparently the solid solubility of zirconium in a highly alloyed nickel-base matrix is at least as great as 0.15%, but its effects on solidus temperatures is considerably less than in unalloyed nickel.

Zirconium additions increase the microporosity levels in the rim sections of the incegral wheels from 0.20% without zirconium to 0.50% at 0.15% zirconium. Zirconium effects were generally consistent with liquid metal pressure considerations; although, the observed increases in microporosity level were generally greater than what the solidification range increases might have otherwise predicted. The number of voids generally increased with zirconium content but the size of the individual voids was unaffected by zirconium content.

Zirconium decreased the 760° C, 60 kg/mm^2 stress-rupture properties in much the same manner as it increased microporosity. The variability between duplicate specimens was much greater with the zirconium additions than with the boron additions. The minimum values, which more appropriately relate to the maximum microporosity measurements, showed a consistent tendency to decrease

Effect of Zirconium on Ni-10Co-9Cr-9W-5.5Al-3.5Ta-1.5Mo-1.5Ti012B									
		Inte	egral Wh	eel Castin	ngs				
Zirco-	Liquidus	Solidus	۸т	Micro-	Stress Rupt 760°C,60kg/mm ²	u.e Life 928°C,30kg/mm ²			
	127290	1205°C	68°C	0.20%	105 hours*	36 hours			
n11 0.07%	1372°C	1300°C	72°C	0.36%	50 hours*	54 hours			
0.15%	1372°C	1288°C	87°C	0.50%	30 hours*	49 hours			

Table 1

* Minimum of two tests

with increasing zirconium content and with higher microporosity levels. However, zirconium seemed to have a beneficial effect, if anything, on the 928°C, 30 kg/mm² rupture properties despite its adverse effect on porosity level. This may be due to its influence on the formation of Ta rich discrete MC particles and retardation of grain boundary film formation.



Fig. 4. Effect of zirconium on primary carbide morphology in Ni-10Co-9Cr-9W-5.5Al-3.5Ta-1.5Mo-1.5Ti.



1000X

Fig. 5. Zirconium-rich segregate in Ni-10Co-9Cr-9W-5.5Al-3.5Ta-1.5Mo-1.5Ti.

0.15Zr



0.15Zr, 0.012B

3000X

Fig. 6. Zirconium-rich "stick phase" in Ni-10Co-9Cr-9W-5.5Al-3.5Ta-1.5Mo-1.5Ti. Zirconium additions had a pronounced effect on the microstructures in addition to its tendency to promote porosity. First of all, zirconium additions tend to alter the morphology of many of the script-like Ta+Ti MC carbides to a more discrete form, Fig. 4. The discrete MC contains about half of the titanium found in the script form. Secondly, the addition of zirconium promotes primary γ' islands - the greater the zirconium, the greater the number and the larger the size of the primary γ' islands. Around many of the primary γ' islands, large areas of a Ni-Zr segregation are present in addition to a boride phase, Fig. 5. Thirdly, the addition of zirconium promotes the formation of a Zr rich "stick phase" thought to be Zr, Ti₂S, C - Fig. 6. Normally, this "stick phase" is found near primary γ' islands, but high zirconium levels cause this phase to form in the grain boundaries as well.

When the boron level is increased to 0.025% at the 0.07% Zr level, the boride phase tends to precipitate out at the grain boundaries in continuous form, Fig. 7. Increasing the Zr to 0.15% tends to suppress this grain boundary reaction.

Boron Effects

The effects of boron were determined at three boron levels: nil, 0.012%, and 0.025% while the zirconium content was held constant at 0.07%. Solidification range, microporosity, and mechanical property measurements at the three boron levels are summarized in Table 2.

Contrary to binary Ni-B phase diagram expectations, the boron additions actually decreased the solidification range of the base alloy from 79°C to 71°C. Apparently the solid solubility of boron in a highly alloyed nickelbase matrix is virtually nil or strong zirconium-boron interactions occur at high boron levels.

The effects of boron on microporosity in the rim sections of the integral wheel castings are shown in Fig. 8, and they were consistent with liquid metal pressure considerations in that microporosity decreased from 0.44% at



0.07Zr, 0.025B 1500X Fig. 7. Continuous boride grain boundary film.

Table 2

				Micro-	Stress Rupture Life	
Boron	Liquidus	Solidus	<u>AT</u>	porosity	760°C,60kg/mm ²	928°C, 30kg/mm2
nil	1380°C	1301°C	79°C	0.44%	1 hour*	14 hours
0.025%	1371°C	1300°C	71°C	-	128 hours*	46 hours

Effect of Boron on Ni=10Co-9Cr-9W-5.5Al-3.5Ta-1.5Mo-1.5Ti-.07Zr Integral Wheel Castings

* Minimum of two tests

 ΔT = 79°C to 0.36% at ΔT = 72°C. The individual voids tended to be smaller but greater in number as ΔT increased.

The effects of boron on the stress-rupture properties of the base alloy were much greater than might have been expected from microporosity considerations alone. The 760°C, 60 kg/mm² ksi rupture life increased from 1 hour without boron to 138 hours at 0.025% boron, while the 928°C, 30 kg/mm² rupture life increased from 14 hours without boron to 54 hours at 0.012% boron. These solute effects on mechanical properties are clearly significant and mandate the addition of trace amounts of boron to nickel-base superalloys regardless of solidification range or microporosity considerations.

In the absence of boron, a fine precipitation at the grain boundary is the MC phase. When boron is added, the majority of the fine grain boundary precipitates are an M_3B_2 boride phase. At the 0.012% B level, large borides are found around primary γ' islands. As the boron is increased to 0.025%, the borides change from the normal script shape to more massive form and the majority of the boride that forms in the boundaries tends to be continuous in nature. The addition of 0.15% Zr suppresses the amount and morphology of the grain boundary boride phase.



nil B, 0.072r 0.44% voids 0.012B, 0.072r 0.36% voids Fig. 8. Maximum microporosity in the rim section at various boron levels. The massive boride phase contains high Cr with lesser amounts of Mo + W. The finer boride phase forming with the Zr rich segregation appears to be even more heavily enriched in Cr.

The relatively constant solidus temperatures coupled with decreasing liquidus temperatures suggest that the boride phases are precipitating directly from the liquid phase during the latter stages of solidification.

Discussion

It has been concluded from basic fluid flow and heat transfer considerations that the propensity to form microvoids in a casting increases with the square of the solidification range of the alloy being cast. The experimental data obtained from large nickel-base superalloy investment castings used in this study generally support this conclusion. Graphic correlations between microporosity and alloy solidification range, such as the log-log correlation illustrated in Fig. 9, show considerable scatter but agree reasonably well with a square relationship with minimum microporosity levels being achieved in alloys with minimum solidification ranges. From binary phase diagram considerations it was hypothesized that significant changes or reductions in the solidification range of nickel-base superalloys might be achieved by simply lowering boron and zirconium at the trace addition level (i.e. <.01%). The results of this study reasonably substantiate this hypothesis in the case of zirconium. Boron significantly affected the solidification range of nickel-base superalloys but in a manner quite opposite from that anticipated. In effect, trace additions of boron reduced the solidification range while trace additions of zirconium increased the solidification range. These opposing effects are probably related to the manner with which these trace solutes partition and precipitate during solidification.

Many highly alloyed nickel-base superalloys precipitate primary γ' directly from the liquid during solidification. The solidus temperature of such alloys corresponds to the γ -primary γ' eutectic temperature. Zirconium lowers the solidus temperature of highly alloyed nickel-base superalloys but does so in a manner different from that expected from simple binary phase diagram considerations. Apparently zirconium lowers the γ -primary γ' eutectic temperature rather than the solidus temperature per se. Detailed metallographic analyses conclusively show that zirconium preferentially partitions





from the primary γ' structures as they form and forms an Ni-Zr intermetallic phase. Such zirconium rich areas, upon a high temperature heat treatment, will react to form ZrC and holes. This phenomena has been observed in less highly alloyed superalloys.

Trace amounts of boron additions up to 0.012% combine with Cr, W, and Mo to form discrete borides in the matrix and in the grain boundaries. Additions up to 0.025% boron promotes continuous grain-boundary borides during solidification. The temperature(s) at which these boride particles precipitate is not at all certain. However, the morphology and location of the borides coupled with the observed liquidus depression suggest that the boride phase is precipitating directly from the enriched liquid during the latter stages of solidification. Apparently boron is moderately soluble in the liquid phase and depresses the liquidus temperature accordingly. Boron precipitates in advance of eutectic solidification and has a nil effect on the solidus temperature.

Boron and zirconium are routinely added to cast nickel-base superalloys as it has been shown that trace amounts enhance the creep-rupture properties of nickel-base superalloys.⁽⁷⁾ The stress-rupture properties measured in the present investigation verify the criticality of boron to the stress-rupture properties and suggests still another advantage of trace boron additionsreduction in microporosity. The detrimental effects of zirconium on increased porosity and microsegregation as found in this study seriously challenges the traditional additions of zirconium to highly alloyed cast nickel-base superalloys.

The present interest in HIPing of castings to close the porosity formed during solidification indicates the relative importance of mechanical properties and porosity. Minimizing the amount of porosity and microsegregation should allow lower HIP temperatures and shorter HIP times to produce sound castings.

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